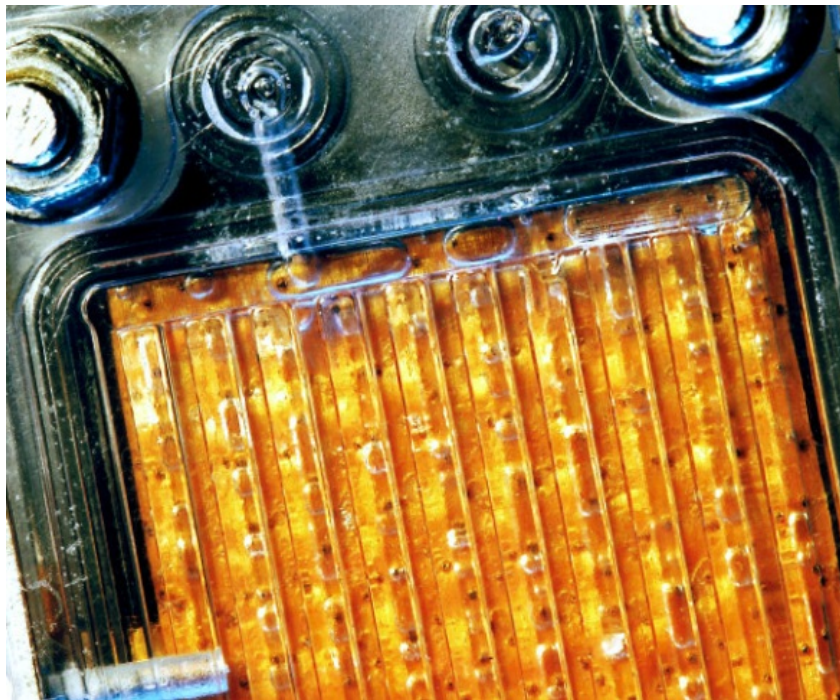

HYDROGEN PRODUCTION BY WATER ELECTROLYSIS: PROGRESS AND CHALLENGES OF A KEY TECHNOLOGY FOR RENEWABLE ENERGIES

A short introduction



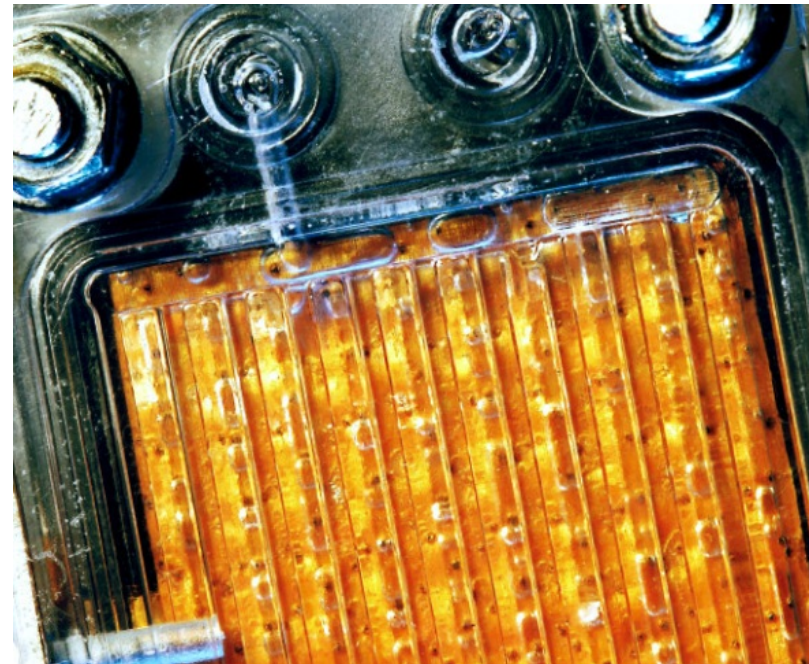
Tom Smolinka

Fraunhofer-Institut für Solare
Energiesysteme ISE

WHTC 2013
Session A3.1: Water Electrolysis
Shanghai, P.R. China, 2013-09-27

Agenda

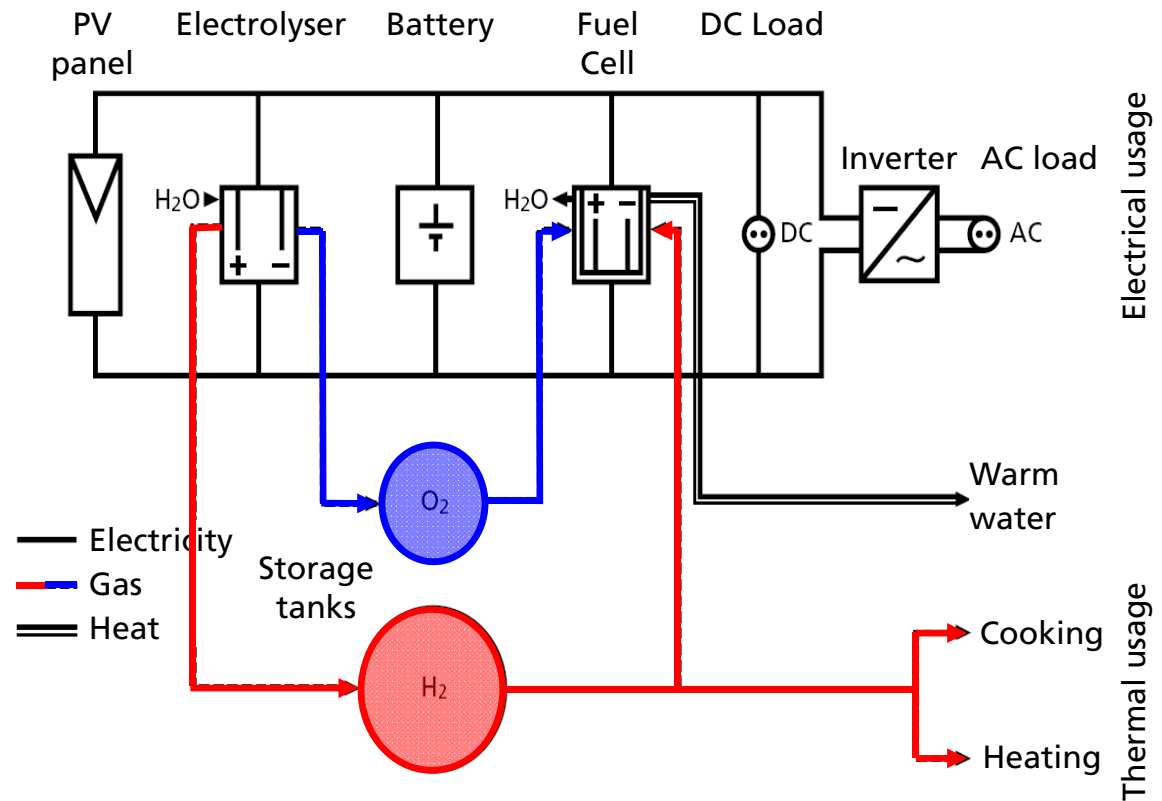
- Water Electrolysis at Fraunhofer ISE
- Typical applications for water electrolysis
- Technical comparison of different types
 - Performance
 - Power range
 - Spec. Energy Consumption
- R&D demand for water electrolysis



Hydrogen Technology at Fraunhofer ISE

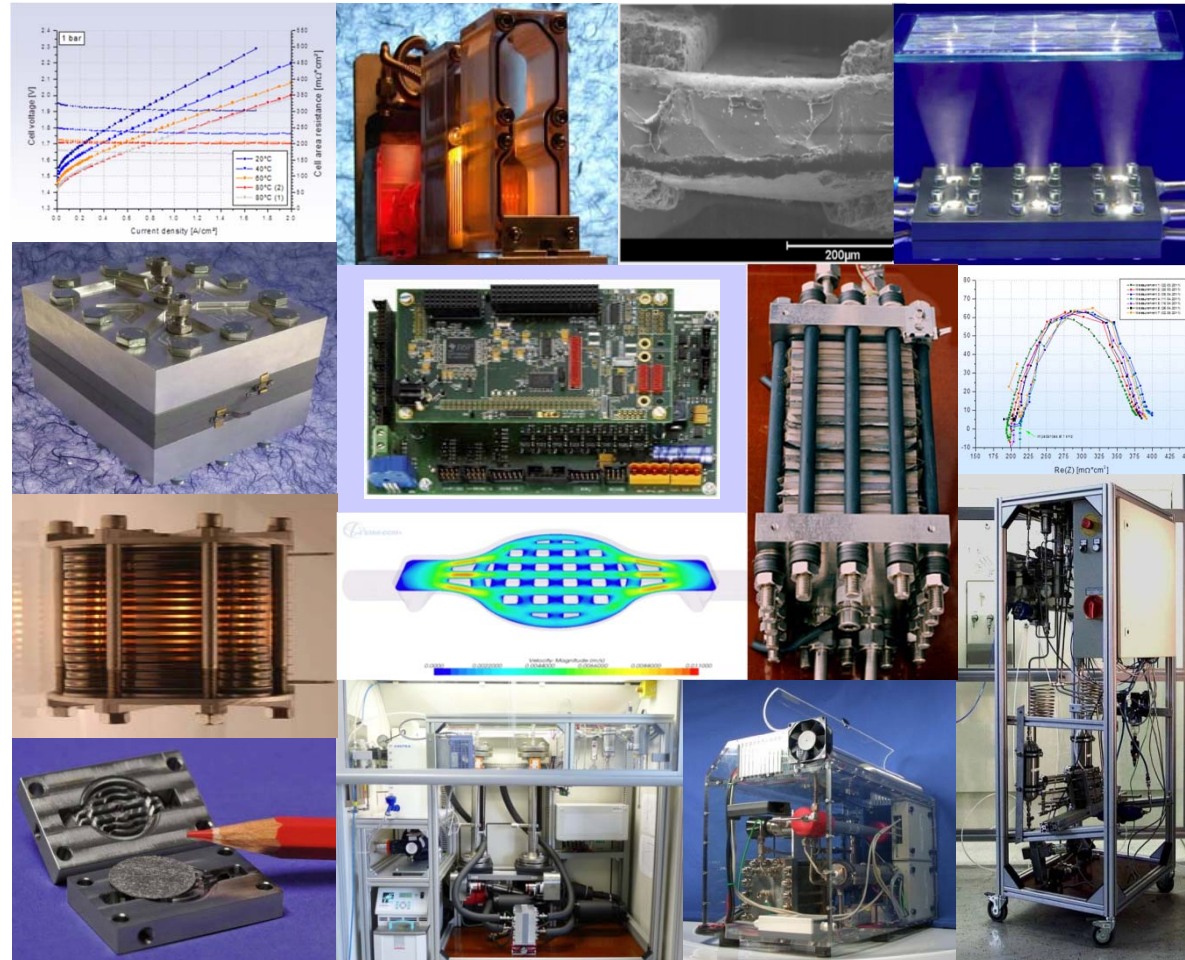
The Self-sufficient Solar House in Freiburg/Germany

- Hydrogen as long-term storage option
- Regenerative fuel cell with PEM technology (EL + storage + FC)
- Field test in 1992-1995



For Nearly 25 Years Experience in PEM Water Electrolysis

- Material characterisation
- Cell and stack design by CFD simulation
- Electrochemical modelling
- Balance of plant
- Control strategies
- Power electronics
- Integration with RES
- System evaluation

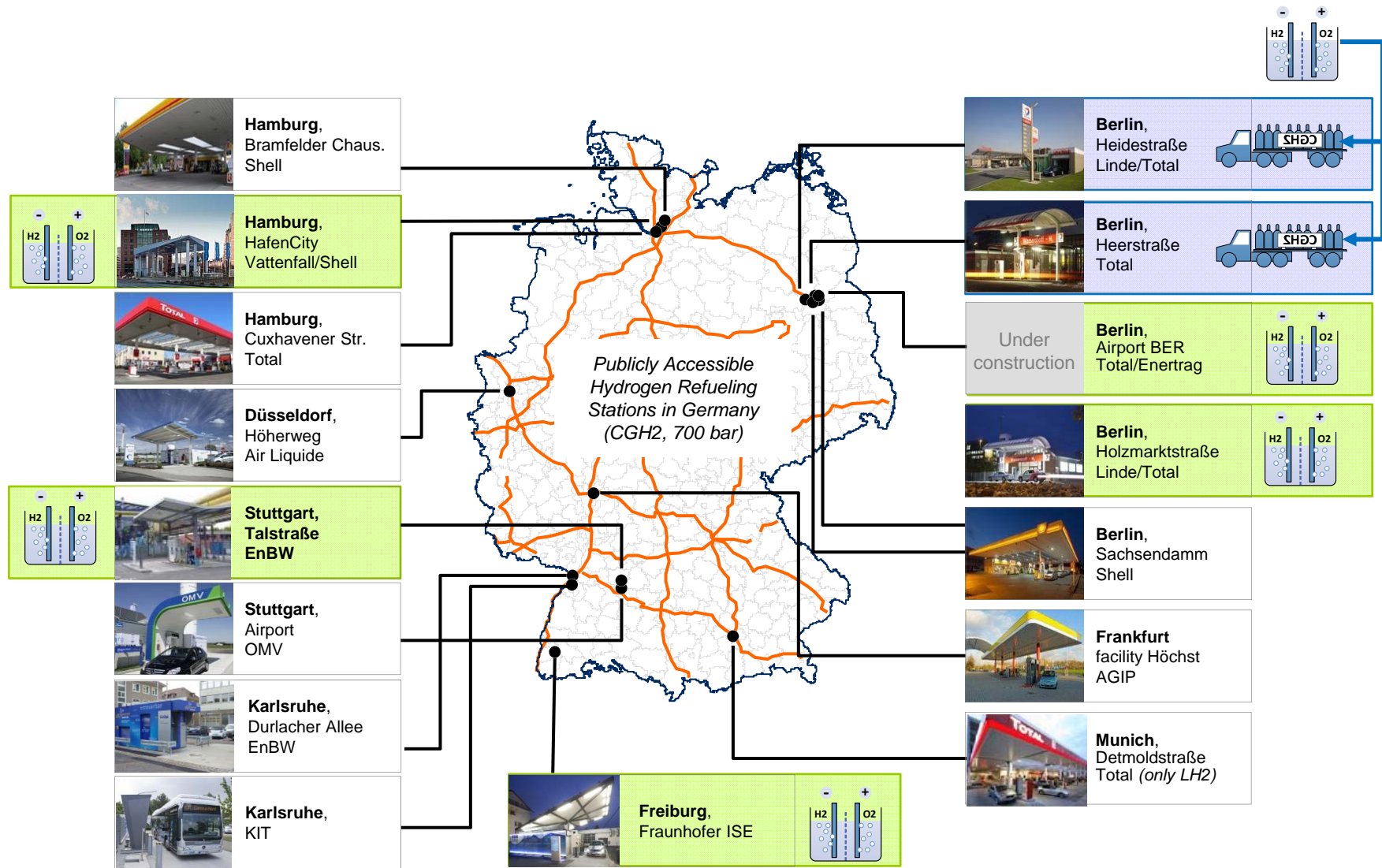


Hydrogen Filling Station at Fraunhofer ISE



- Main components of the filling station:
 - (Pressure) electrolyser (30 bar / 6 Nm³/h)
 - Mechanical compressor
 - Storage tanks
 - Dispenser units (200/350/700bar)
 - Filling according to SAE J2600
- Integrated container solution
- Publicly accessible filling station
- Located at premises of Fraunhofer ISE
- Coupled with renewable energies:
 - Photovoltaic modules (roof)
 - Certified green electricity

700bar Hydrogen Refueling Stations in Germany



Project "Power to Gas für Hamburg"

Partners:



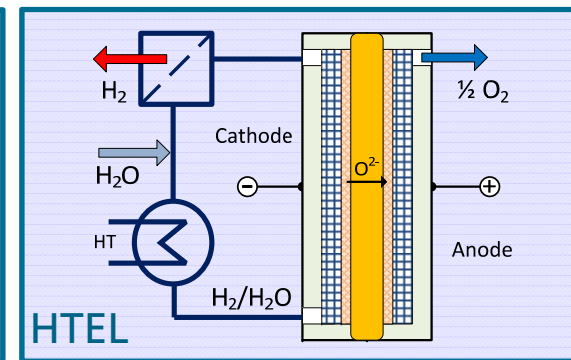
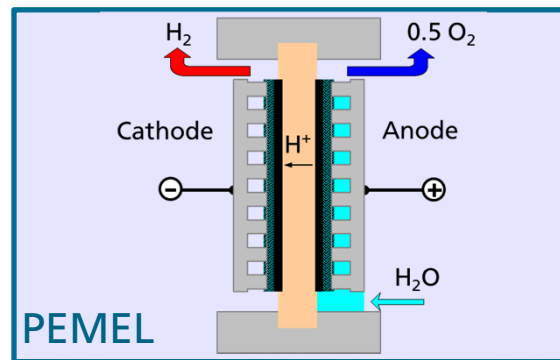
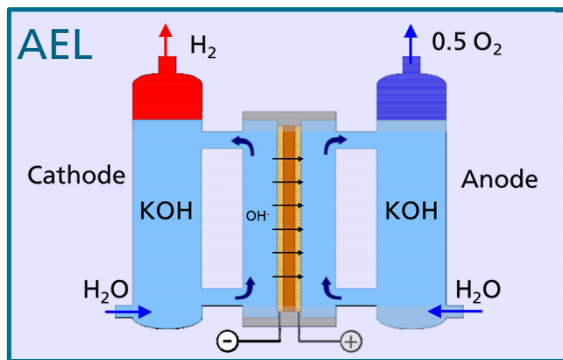
Facts:

- Run time: 3 years, 11/2012 - 10/2015
- PEMEL in one container
- 1 MW_{el} single stack!
- Direct feeding into the NG grid
- Hydrogen concentration will be < 1 vol%

Electrolytical Water Splitting:



Technology	Temp. Range	Cathodic Reaction (HER)	Charge Carrier	Anodic Reaction (OER)
Alkaline electrolysis	40 - 90 °C	$2\text{H}_2\text{O} + 2e^- \Rightarrow \text{H}_2 + 2\text{OH}^-$	OH^-	$2\text{OH}^- \Rightarrow \frac{1}{2}\text{O}_2 + \text{H}_2\text{O} + 2e^-$
Membrane electrolysis	20 - 100 °C	$2\text{H}^+ + 2e^- \Rightarrow \text{H}_2$	H^+	$\text{H}_2\text{O} \Rightarrow \frac{1}{2}\text{O}_2 + 2\text{H}^+ + 2e^-$
High temp. electrolysis	700 - 1000 °C	$\text{H}_2\text{O} + 2e^- \Rightarrow \text{H}_2 + \text{O}^{2-}$	O^{2-}	$\text{O}^{2-} \Rightarrow \frac{1}{2}\text{O}_2 + 2e^-$



Hydrogen Production by Water Electrolysis

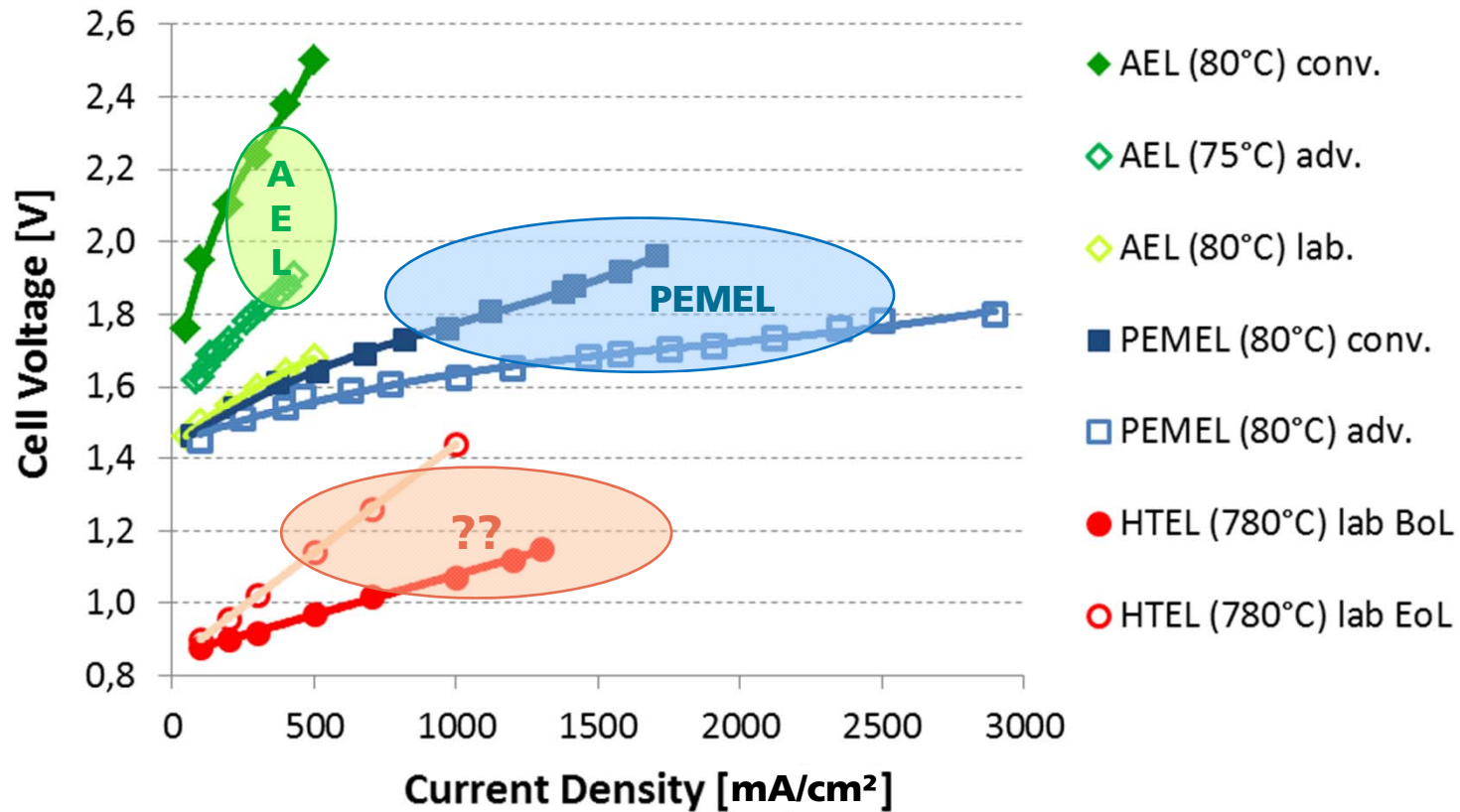
Overview

	Alkaline Electrolysis	Membrane Electrolysis	Solid Oxide Electrolysis
Electrolyte	Liquid alkaline KOH	Solid acid polymer	Ceramic metal compound
Charge carrier	OH ⁻	H ⁺	O ²⁻
Electrodes	Ni/Fe electrodes (Raney)	Noble metals (Pt, Ir, ..)	Ni doped ceramic
Temperature	50-80 °C	RT - 90 °C	700 - 1,000 °C
Pressure	< 30 bar	< 200 bar	Atm
Modul size	Max. 760 Nm ³ H ₂ /h ~ 3.2 MW _{el}	Max. 30 Nm ³ H ₂ /h ~ 170 kW _{el}	~ 1 Nm ³ H ₂ /h kW range



Comparison of Performance Data

Vi Curves for Different EL Technologies



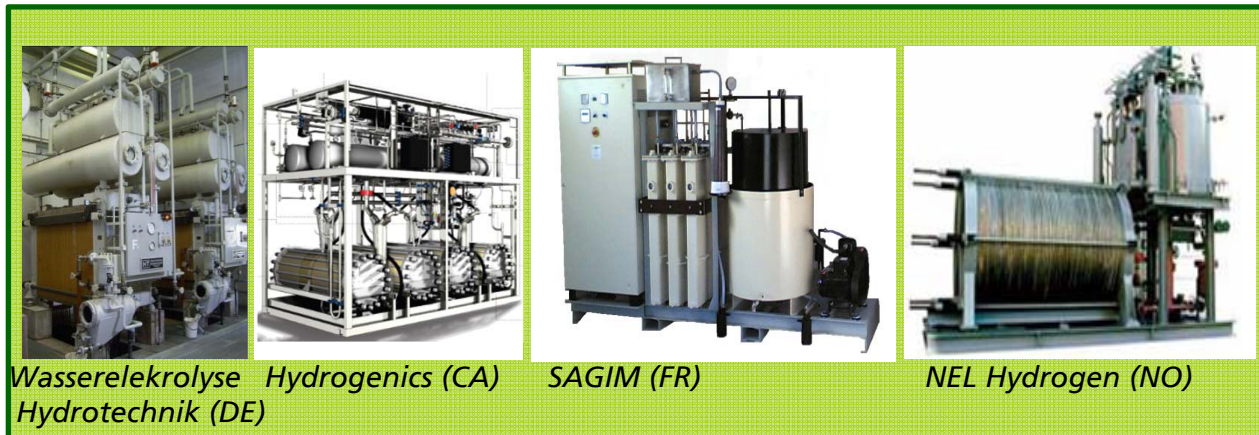
Commercial Available Electrolysis Systems

- Alkaline electrolysis

- 1 - 760 Nm³/h
- 5 kW_{el} - 3.4 MW_{el}

- PEM electrolysis

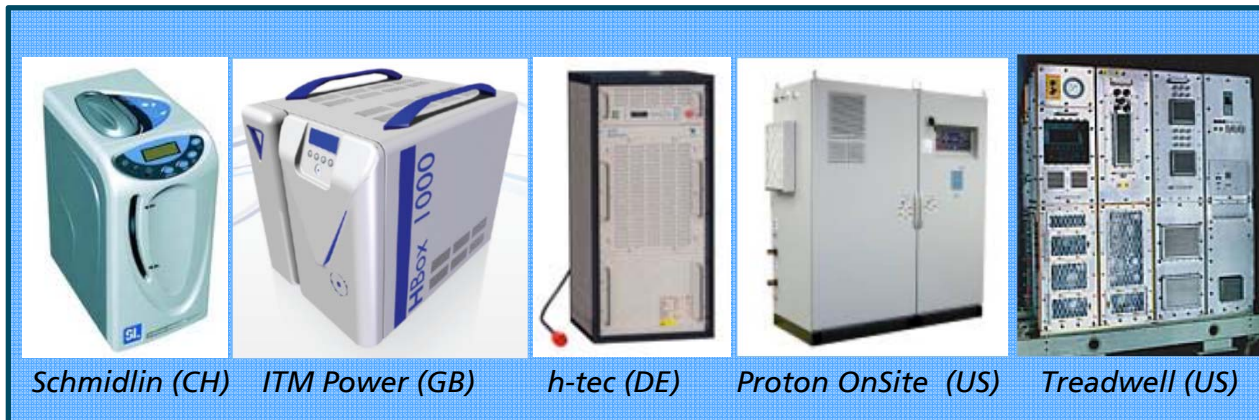
- 0.01 - 30 Nm³/h
- 0.5 - 160 kW_{el}



- PEMEL grows!

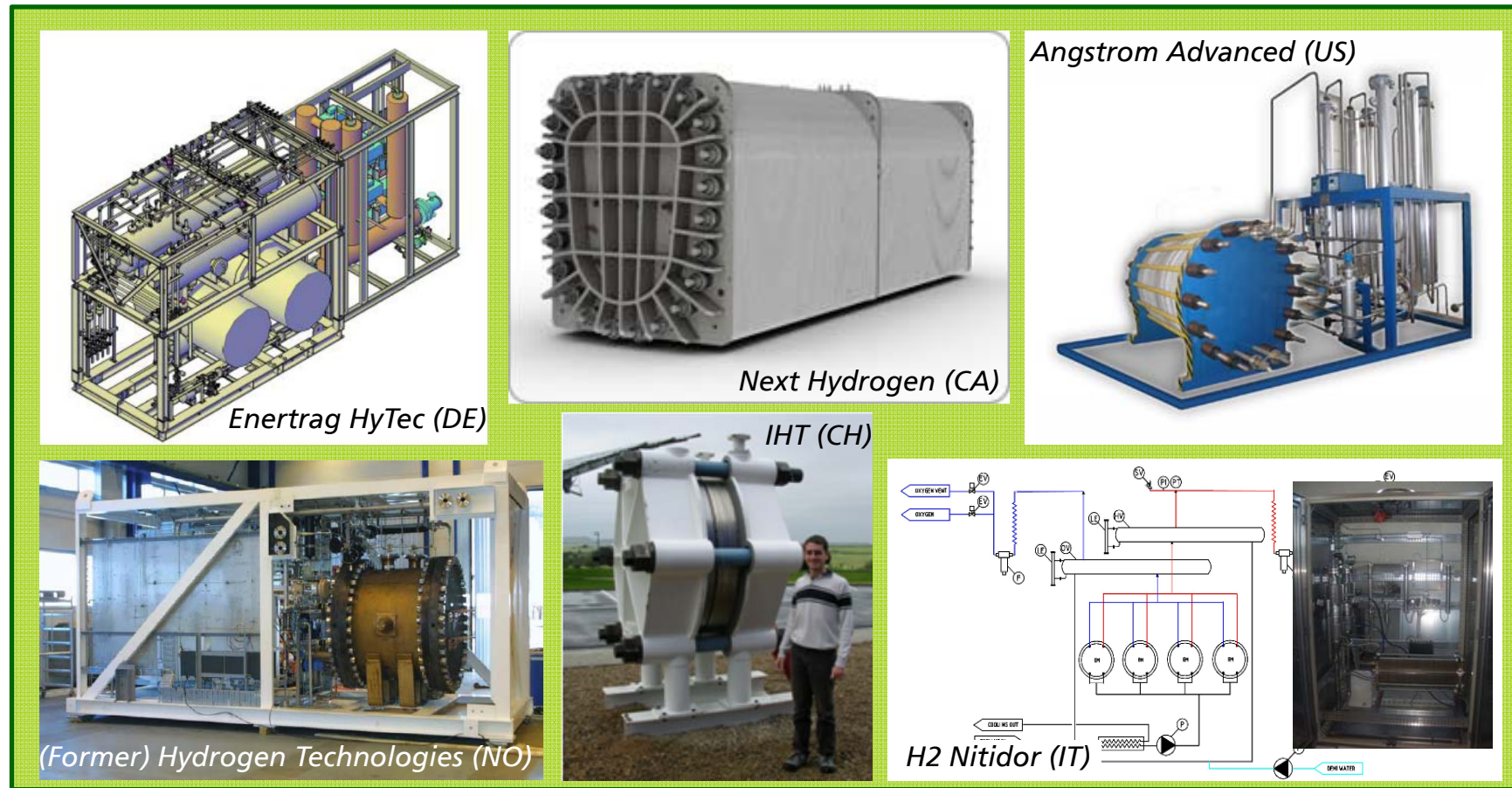
- In 2 - 3 years:

- Up to 250 Nm³/h
- Up to 1.0 MW_{el}



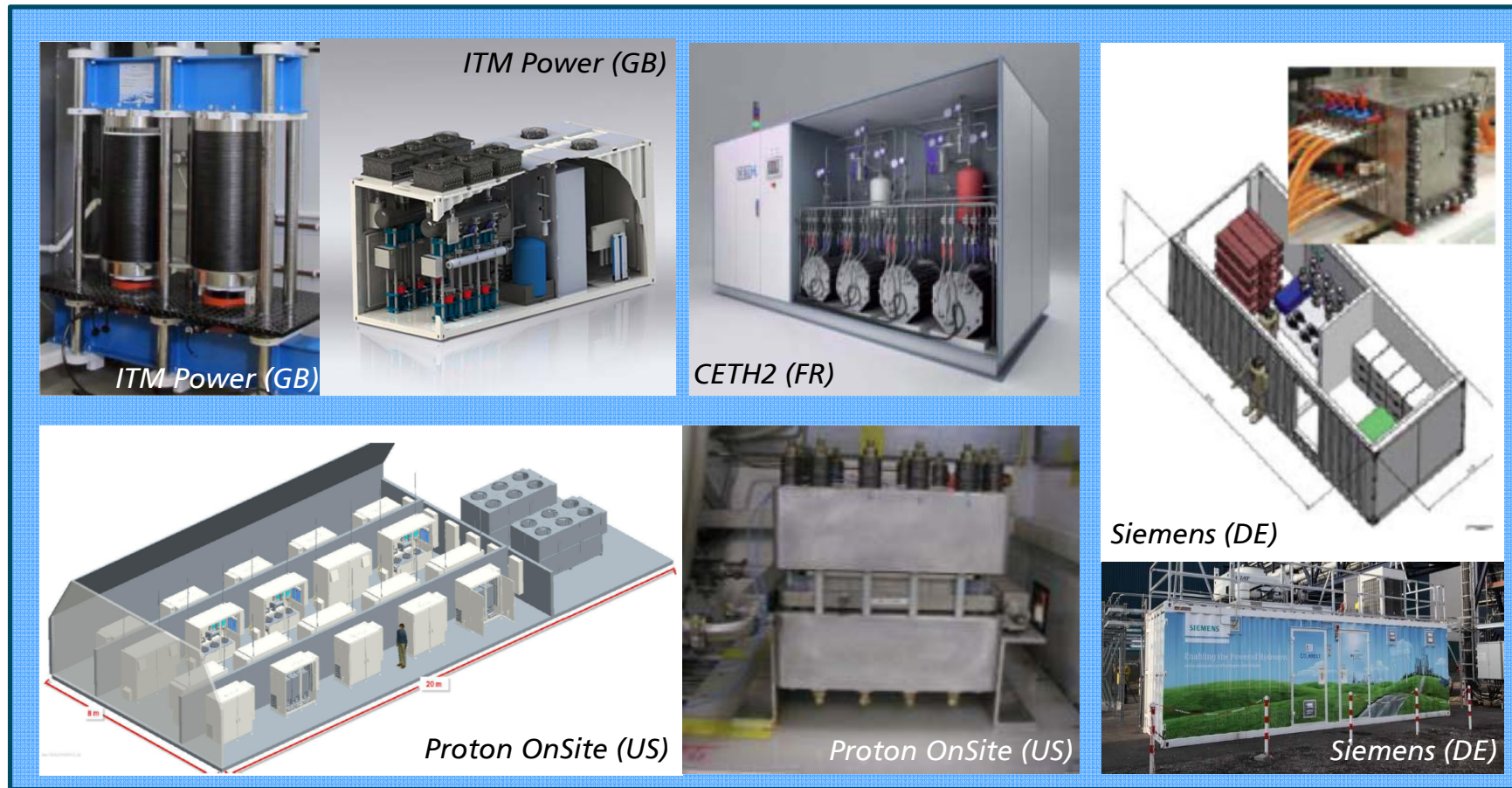
Development Trends in Water Electrolysis

Pressurised Alkaline Electrolysers (Re)Enter the MW Class



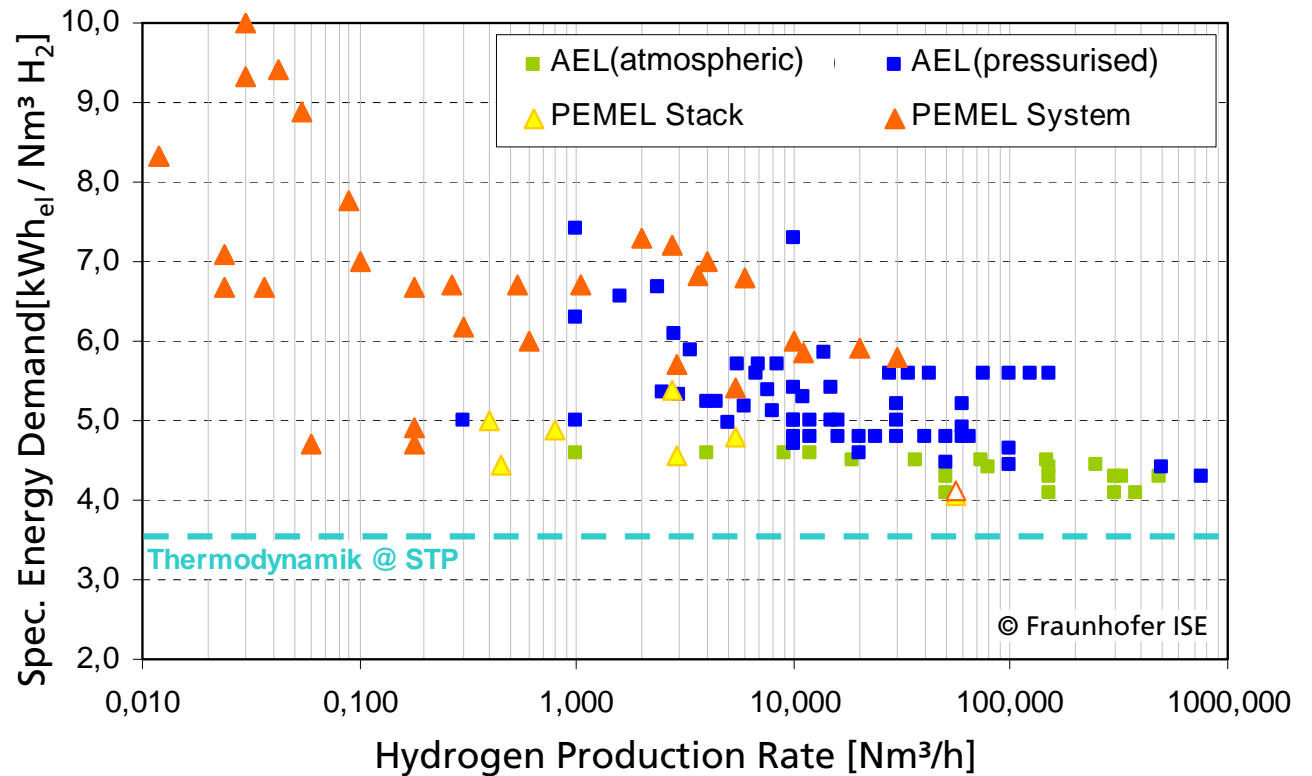
Development trends in Water Electrolysis

PEM Electrolysers Entering MW Class as well.



Specific Energy Consumption: Attempt to Compare the Efficiency of Electrolysers

- Manufacturer's data
- No standardised data
- Different pressure and H₂ purity
- Specifications for steady state operation



Where Do We Have R&D Demand in the Next Years?

■ AEL

- Increasing current density
- (Increasing pressure tightness)
- Faster dynamics of the complete system (BOP)
- Higher part load range
- Decreasing production costs through economies of scale

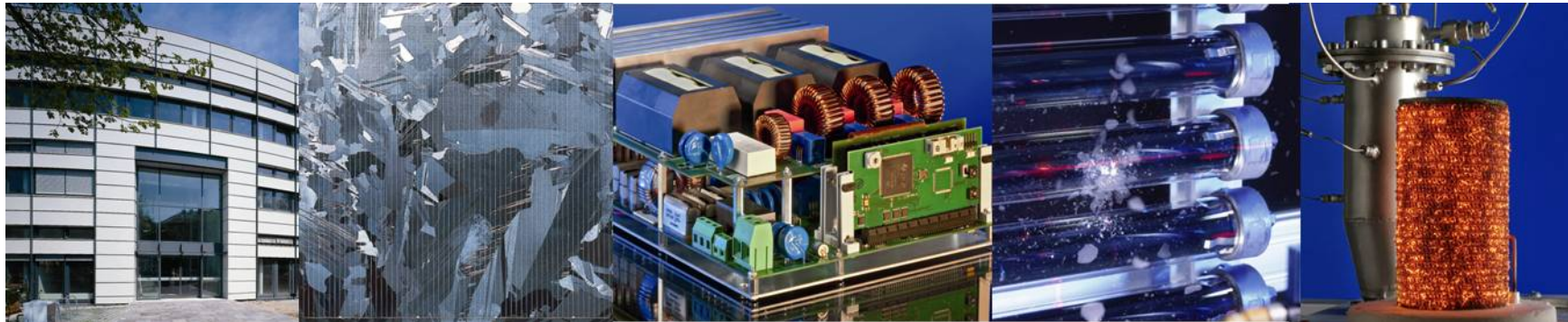
■ PEMEL

- Increasing life time of materials/ stack
- Scale up concepts for stack and system
- Decreasing costs by substitution or reduction of expensive materials
- (Decreasing production costs through economies of scale)

■ HTEL

- Development of adapted electrodes/ electrolyte for SOEL
- Cell and stack design
- Proof of life time
- Pressure tightness
- Cycling stability

Thanks a lot for your kind attention!



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